



is lost as waste heat.

When the process needs more heat, all of the steam is sent to the process heater. Steam passes through the condenser

( $\dot{m}_6 = 0$ ) and waste heat losses do not occur ( $Q_{\text{waste heat}} = 0$ ). If this is not enough, some of the vapor formed in the boiler is directed to the process heater via a throttling valve or a pressure reducing valve. The highest value of the process heat is obtained by passing all the vapor obtained

from the boiler to the pressure reducing valve ( $\dot{m}_{1,2,3} = \dot{m}_4$ ). Power generation does not happen in this case. If the process does not have a heat load, all the steam is passed through the turbine and the condenser ( $\dot{m}_4 = \dot{m}_5 = 0$ ). The system works like a steam power cycle.

In this article, solid wastes different thermal properties obtained from the Aladağ district of Bartın province were used as the working fluid as the heat source of the CHP system and energy analysis were made by using 6 different fluids in the isentropic, wet and dry properties as the working fluid (R600, Isopentane, R141b, R123, R134a, R32) in order to find the optimum fluid for the system. In addition, steam turbine (Organic Rankine Cycle) is used for power generation in the system. Cooling fluids with a flow rate of 5 kg/s were used in the system and the calculations were taken into account when 10% of the steam obtained from the boiler was expanded before the turbine and 70% was expanded to a predetermined pressure of 500 kPa in the turbine before being sent to the process heater.

## B. Working fluid selection for system

One of the important variables in thermodynamic systems is the determination of the refrigerant that provides energy transfer in the system. The determination of the fluid is to simulate different fluids for the cycle by some factors. A total of 6 different refrigerant fluids (R600, Isopentane, R141b, R123, R134a, R32) were selected for dry, isentropic, wet properties to be used as working fluid in the system. Parameters such as fluid availability, chemical stability, safety, toxicity, flammability, ecological effects (global warming potential GWP, ozone depletion potential ODP) were taken into account in selecting the fluids and their requirements coverage rates were examined.

As a result of the examination, the optimum fluid with the most suitable features for the system was determined. The thermodynamic parameters obtained and used in the study are obtained from the Engineering Equation Solver (ESS).

The properties of the refrigerants used are shown in Table 1.

**Table 1.** The properties of the refrigerants [5]

Property	Component	Molar mass (g/mol)	Critical temperature (°C)	Critical pressure (MPa)	ODP	GWP (100year)
Dry	R600	58.12	152.0	3.80	0	~20
Dry	Isopentane	72.15	187.2	3.38	0	~20
Isentropic	R141b	116.95	204.4	4.21	0.120	717
Isentropic	R123	152.93	183.7	3.66	0.010	77
Wet	R134a	102.03	101.1	4.06	0	1370
Wet	R32	52.02	78.1	5.78	0	716

## C. Thermodynamic equations for the energy transfer

All of the equipment forming the cogeneration system using the Organic Rankine Cycle includes continuous flow-open thermodynamic processes. In continuous flow-open systems, heat and work transition between environment and system do not change with time. At the same time, all fluid properties at the inlet and outlet of the control volume are stable.

Mass conservation in continuous flow-open systems;

$$\sum \dot{m}_{in} = \sum \dot{m}_{out} \quad (1)$$

As the energy change in the control volume is constant

( $\Delta E_{Cv} = 0$ ), the amount of energy (mass, heat, work) entering the system at that time should be equal to the amount of energy generated.

Energy conservation in continuous flow-open systems;

$$\dot{E}_{in} - \dot{E}_{out} = \frac{dE_{sys}}{dt} = 0 \quad (2)$$

$$\dot{E}_{in} = \dot{E}_{out} \quad (3)$$

Energy from the unit mass of the fluid;

$$E_{mass} = (u + PV) + ke + pe = h + ke + pe \quad (4)$$

if it is written in the equation;

$$\dot{Q}_{in} + \dot{W}_{in} + \sum_{in} \dot{m}E = \dot{Q}_{out} + \dot{W}_{out} + \sum_{out} \dot{m}E \quad (5)$$

$$\dot{Q}_{in} + \dot{W}_{in} + \dot{m}_{in}(h + V^2/2 + gz) = \dot{Q}_{out} + \dot{W}_{out} + \dot{m}_{out}(h + V^2/2 + gz) \quad (6)$$

$$\dot{Q} - \dot{W} = \dot{m}(h_{out} - h_{in} + \frac{V_{out}^2 - V_{in}^2}{2} + gz_{out} - gz_{in}) \quad (7)$$

Kinetic and potential energy changes in the system are often neglected because they are small compared to heat and work transitions. ( $\Delta ke \cong 0, \Delta pe \cong 0$ )

$$\dot{Q} - \dot{W} = \dot{m}(h_{out} - h_{in}) \quad (8)$$

A part of the heat energy entering the heat machines can be transformed into work. The part of the thermal energy entering the system that can be converted to net work is a parameter of system efficiency and is called thermal efficiency ( $\eta_{th}$ ).

The desired result in the heat engines is the net work, the required value, the net amount of heat given to the system working fluid.

The thermal efficiency defined as;

$$\text{thermal efficiency} = \frac{\text{net work obtained}}{\text{total incoming heat}} \quad (9)$$

$$\eta_{th} = \frac{\dot{W}_{net}}{\dot{Q}_{incomin g heat}} \quad (10)$$

**D.Conditions for the CHP Calculation**

**Table 2.** Input parameters to the system

Parameters	Values
Inlet temperature of heat source	150 °C.
Ambient temperature	25 °C.
Cooling water temperature at the condenser	25 °C
Mass flow of the heating fluid (Therminol55 )	10 kg/s
Isentropic efficiency of the pump1,2	0.85
Isentropic efficiency of the turbine	0.85
Efficiency of the boiler	0.95
Mass flow of the working fluid	5 kg/s
Isobaric heat capacity (Therminol55)	2.17 kJ/kg.K
Isobaric heat capacity cooling water	4.179 kJ/kg.K
Boiler pressure	10 MPa
Interval pressure	500 kPa
Condenser pressure	10 kPa
System runtime	12 hour

**RESULTS AND DISCUSSION**

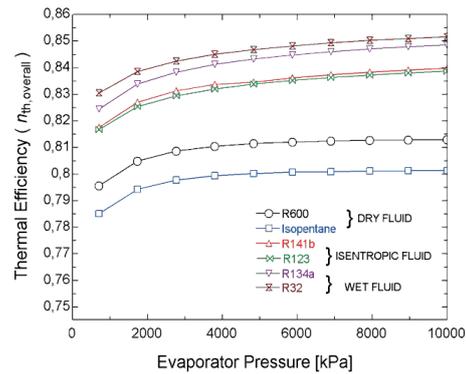
The Engineering Equation Solver (EES), which is useful when analyzing the system, is taken from the written program. The parameters used in the cycle energy analysis are shown in Table 2.

Figure 2 shows the effect of the evaporator pressure on the overall thermal efficiency for system. As a result of the comparative analysis, the efficiency of dry (R600-Isopentane) fluids were lower than those of isentropic (R141b-R123) and wet fluids (R134a-R32) at the same evaporator pressure values.

When the evaporator pressure increases from 700kPa to 10MPa this ratio increased from 79.56% to 81.9% for R600a and from 78.52% to 80.13% for Isopentane.

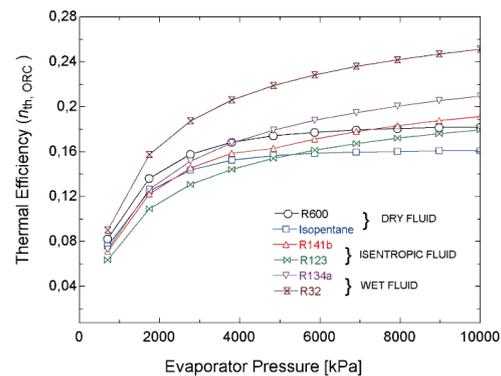
The lowest values were observed for isentropic fluids at the same pressures after the dry fluids. As the pressure increased, efficiency of 141b and R123 increased by 2.22% and 2.19%, respectively and it was determined that the values and changes of two fluids are very close to each other under the same conditions.

The highest efficiency values were seen in wet liquids. The efficiency at maximum pressure is 84.86% at R134a and 85.17% at R32.



**Figure 2.** Exchange of overall thermal efficiency with Evaporator Pressure

Figure 3 shows the effect of the evaporator pressure on the thermal efficiency of ORC. The increase in pressure boosted the ORC efficiency and this increase was read at least in Isopentane (7.115% - 16.09%) and maximum in R32 (9.02% - 25.12%). The highest efficiency values were recorded at R32 (9.02% at 700 kPa, 25.12% at 10 MPa).



**Figure 3.**Exchange of ORC thermal efficiency with Evaporator Pressure

**CONCLUSION**

In this study, the energy analysis of the combined heat power systems (CHP) was carried out using EES. To evaluate the CHP program was written at EES.

The increase in evaporative pressure has increased system efficiency. This increase has more impact on overall efficiency than ORC efficiency.

The high overall efficiency values of the system have shown that the use of cogeneration systems provides a better utilization of energy than systems where heat and power are produced separately.

The use of solid waste fueled cogeneration system in this study is of great importance for clean and efficient energy production.

The cooling fluid to be used is selected for the system to work in optimum conditions. It has been observed that under the same conditions, the R32 refrigerant is advantageous over other refrigerants in terms of both ORC and overall

thermal efficiency.

The resulting data and fluid properties (such as chemical, physical, biological properties) have shown that R32 fluid may be an important alternative to clean, sustainable energy in the future at the same time.

### Acknowledgment

This present work was developed within the the framework of a research project having ID 2018-FEN-A-014 fully funded by Bartın University, Bartın University Scientific Research Projects Coordinator ( BAP ) . The authors would like to thank Bartın University , BAP for the financial support given to the project.

Nomenclature	
$c_p$ isobaric heat capacity, kJ/kg.K	$\eta_{turbine}$ isentropic efficiency of the turbine
$h$ specific enthalpy, kJ/kg	$V$ specific speed , m/s
$\dot{m}$ mass flow rate, kg/s	$g$ acceleration of gravity, $m/s^2$
$P$ pressure, kPa	$z$ reference height , m
$s$ specific entropy, kJ/kg.K	$\eta_{th}$ thermal efficiency
$T$ temperature, °C, K	$t$ time, s
$\dot{Q}$ heat input, kW	<b>Subscripts</b>
$\dot{W}$ work, kW	1,2,3.....11 state point
$e_e$ energy utilization rate	$ke$ kinetic energy, kJ
$\eta$ efficiency	$pe$ potential energy ,kJ
$\eta_{pump,1,2}$ isentropic efficiency of the pump	$in$ inlet
$\eta_{boiler}$ efficiency of the boiler	$out$ outlet

## REFERENCES

- [1] Alexander Zerrahn ,Wolf-Peter Schill , Claudia Kemfert ,On the economics of electrical storage for variable renewable energy sources. European Economic Review 2018;108:259–279.
- [2] Simon Jenniches, Assessing the regional economic impacts of renewable energy sources – A literature review. Renewable and Sustainable Energy Reviews 2018;93:35–51.
- [3] Gerbaulet C, Egerer J, Oei PY, Paeper J, Hirschhausen C. Die Zukunft der Braunkohle in Deutschland im Rahmen der Energiewende. Berlin: DIW; 2012.
- [4] Anders N. Andersen ,Henrik Lund ,New CHP partnerships offering balancing of fluctuating renewable electricity productions. Journal of Cleaner Production 2007;15: 288-293
- [5] J.M Calm,G.C. Hourahan,ICR 2011,August 21-26-Prague,Czech Republic